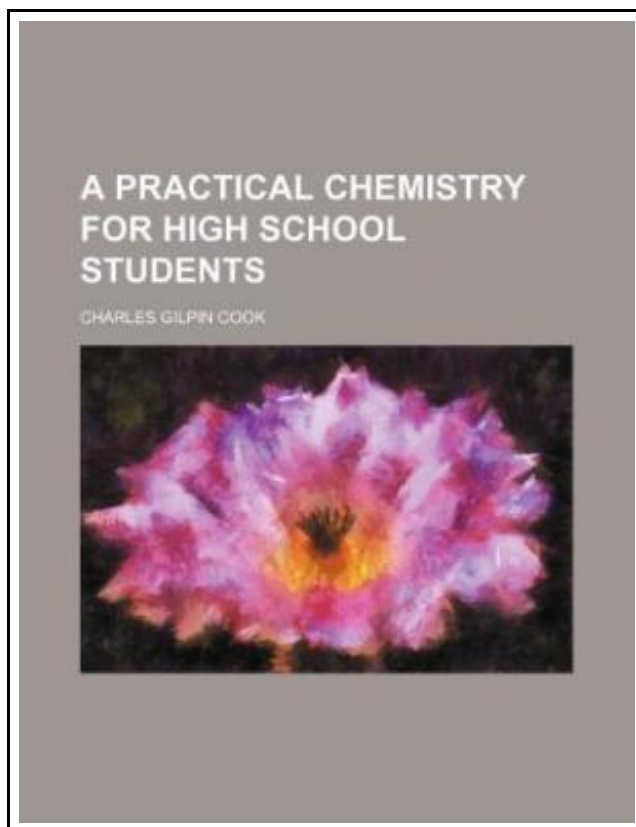


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(Lexie Paucek PhD)

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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1916 Excerpt: . nascent condition: $H_2O_2 = H_2O + O$. In the presence of any substance capable of being easily oxidized this reaction is quite rapid. All are familiar with the frothing produced when hydrogen peroxide is poured upon a sore. Hydrogen peroxide is sometimes sold under the name of dioxygen. Formaldehyde.--A very different though equally active antiseptic is formaldehyde. Its formula is $HCHO$. It is the simplest of a number of aldehydes, which are compounds formed by the oxidation of the alcohols. Formaldehyde is made by the oxidation of methyl, or wood alcohol, CH_3OH . A 40 per cent. solution of formaldehyde gas in water is used as a disinfectant of rooms where there has been a contagious disease. This is conveniently done by pouring the solution on quicklime. The reaction between the lime and the water produces sufficient heat to vaporize the formaldehyde, which penetrates to every part of the room and kills the bacteria. The chemical action of this disinfectant is that of a reducing agent, that is, it abstracts oxygen from other things and is oxidized to formic acid. Carbolic Acid.--For general purposes of disinfecting probably nothing is better than carbolic acid, since it can be used in such a variety of ways. The quantity required, however, is greater than of most antiseptics. The term carbolic acid is usually applied to a water solution of phenol, though sometimes phenol is called carbolic acid. Phenol, C_6H_6O , is a white crystalline substance with a characteristic odor. It..



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